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4.2 For some
hypothetical metal, the
equilibrium number of
vacancies at 900 C is
 $2.3 \times 10^{25} \text{ m}^{-3}$. If the
density and atomic
weight of this metal
are 7.40 g/cm^3 and
 85.5 g/mol ,
respectively, calculate
the fraction of

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vacancies for this
metal at 900 C.

Solution This problem
is solved using two
steps: (1) calculate the
total number of lattice
sites in silver, N , using

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